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## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

A01N 25/00, 55/02, 59/16, 59/20  A1  (43) International Publication Date: 26 October 2000 (26.  (24) International Application Number: PCT/IL00/00219  (25) International Filing Date: 13 April 2000 (13,04,00)  (26) Priority Data: 15 April 1999 (15,04,99)  (27) International Filing Date: 13 April 2000 (13,04,00)  (38) Priority Data: 09/29],312  (39) Priority Data: 15 April 1999 (15,04,99)  (71) Applicant (for all designated States except US): AGRICARE LTD. [IL/IL]; 39 Aharonson Street, 52293 Ramat Gan (IL).  (72) Inventor; and (75) Inventor/Applicant (for US only): TSIVION, Yoram [IL/IL]; 3 Zeller, Street, 46725 Herzlia (IL).  (74) Agent: EITAN, PEARL, LATZER & COHEN-ZEDEK; 2 Gav Yam Center, 7 Shenkar Street, 46725 Herzlia (IL).  (54) Title: AGENTS AND METHODS FOR THE CONTROL OF FUNGAL AND BACTERIAL DISEASES  (57) Abstract  A method and agents for the control of fungal and bacterial diseases are provided. In one embodiment, specific chelates of omer metal ions operate to control bacterial caused diseases in plants. In another embodiment, fungal disease in plants is controlle employing a chelate of zinc ion, together with a chelate of another metal. Some of the metals preferred for use in combination with comprise copper and/or managenese. Some preferred chelating agents comprise velvate, citric acid or resominol. The chelates for such as the comprise opper and/or managenese. Some preferred chelating agents comprise velvate, citric acid or resominol. The chelates of another metal.	INTERNATIONAL APPLICATION PUBLISH		
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# AGENTS AND METHODS FOR THE CONTROL OF FUNGAL AND BACTERIAL DISEASES

## FIELD OF THE INVENTION

This invention relates generally to disease control. More specifically, the invention relates to the control of diseases caused by fungi or bacteria. Most specifically, the invention relates to methods and materials for controlling fungal or bacterial disease in plants.

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#### BACKGROUND OF THE INVENTION

Fungal diseases are very common in crops in various stages of the growth cycle, sometimes with devastating effect. Commercially there are many control agents to combat such diseases, belonging to diverse chemical groups.

IL 97676 discloses compositions and a method for the control of fungal diseases using citric acid chelates of copper, zinc, manganese and calcium either singly or in combination. The specific identity of the diseases was not disclosed apart from the fact that the pest was a fungus in each example given.

The Pesticide Manual (Ed. C.R. Worthing, The British Crop Protection Council Ninth Edition, 1991), lists several commercial fungicides which are complexes of thiocarbamates with one of the following metals: manganese, zinc, iron, copper or specific mixtures of such complexes. The carbamates *per se* are well known fungicides and are used in that sense also without the addition of metal.

Bacterial diseases occur in many crops in various stages of the growth cycle, sometimes with devastating effect, such as fireblight in pears and apples.

Some storage or "postharvest" diseases of agricultural products are caused by bacteria.

Commercially there are not many control agents to combat such diseases, the most prominent ones are antibiotics and copper based control agents.

#### SUMMARY OF THE INVENTION

It is an object of the invention to provide a composition for controlling fungal and bacterial diseases in plants.

According to the invention there is provided a composition for the control of fungal diseases which contains at least one type of chelating agent complexed with zinc ion and with at least another metal ion, whereas the chelating agent's molecules are characterized in being either synthetic or natural.

According to the invention, bacterial pathogens can be controlled by treating the infected plants or cultures of the bacteria with chelated metal ions, most notable in this respect is zinc. However, other chelated metals are active singly, or in mixture with chelated metals, notably with chelated zinc.

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A second aspect of the invention is that a certain bacterial pathogen tends to be more susceptible to a specific chelating agent or a combination of such agents in connection with a certain metal or with a combination of such metals, than other bacterial pathogens.

In one specific embodiment, specific chelates of one or more metal ions operate to control bacterial caused diseases in plants. These chelates may be based upon organic acids, organic alcohols or amines, aromatic ring substitution products, or other synthetic chelating agents combined with a metal, which may comprise zinc, copper, or manganese, used either singly or in combination. In another aspect of the present invention, it has been found that fungal disease in plants can be effectively controlled by employing a chelate of zinc ion, together with a chelate of another metal. Some of the metals preferred for use in

combination with zinc comprise copper and/or manganese. Some preferred chelating agents comprise glycine, citric acid or resorcinol.

These agents are highly effective in controlling both bacterial and fungal diseases in plants.

# DETAILED DESCRIPTION OF THE PRESENT INVENTION

#### Fungal infestation

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#### Preparation of Compositions

Following are various compositions (termed henceforth "test compositions") that were tried in various experiments, for their effect on fungal pathogens. All preparatory steps were carried out accompanied by occasional stirring, until a clear solution was obtained in each case.

#### i) Citric Acid Chelates

Composition A: a measured amount of copper (supplied as CuSO<sub>4</sub>·5H<sub>2</sub>O), was thoroughly mixed in water with equimolar amount of citric acid, then water added to form a final 0.83 M copper in solution.

Composition B: a measured amount of manganese (supplied in the form of MnSO<sub>4.1</sub>H<sub>2</sub>O, was thoroughly mixed in water with equimolar amount of citric acid, then water added to form a final 1.65 M manganese in solution.

Composition C: a measured amount of zinc (supplied in the form of ZnSO<sub>4.7</sub>H<sub>2</sub>O), was thoroughly mixed in water with equimolar amount of citric acid, then water added to form a final 1.5 M zinc in solution.

Composition D: an equal volume mixture of composition A and B. Final concentration of total metals: 1.2 M

Composition E: an equal volume mixture of composition A, B and C. Final concentration of total metals: 1.3 M.

Table 1: Colony Size (in cm) After Incubation with Various C ncentrations (% in growth medium) of the Test Compositions.

Test Composition	0 %	1.0 %	2.5 %
Δ	3.5	0.0	0.0
B	3.5	2.0	1.63
D	4.5	2.0	1.2
	0.5	0.0	0.0

In this example, manganese chelate, and mixed copper and manganese chelates were least effective. The copper chelate and the mixed zinc copper and manganese chelates were much more effective.

<u>Example 2</u>. Botrytis cynerea, a common fungal pest, was treated in plates as described above, results are shown in Table 2.

Table 2:. Colony size (in cm) after Incubation with Various Concentrations (% in growth medium) of the Test Compositions.

Test Composition	0 %	1.0 %	2.5 %
Δ	9.0	8.0	0.27
R	9.0	9.0	7.6
Director of the second of the	9.0	9.0	9.0
F	9.0	2.4	1.9

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In this example, as in the former one, manganese chelate, and mixed copper and manganese chelates were least effective. The mixed chelate of zinc, copper and manganese was most potent in the lowest effective concentration (1%), although the copper chelate had a better effect in the higher concentration (5%).

#### ii) Amino Acid Chelates

Composition F: a measured amount of copper (supplied in the form of CuSO<sub>4</sub>·5H<sub>2</sub>O), was thoroughly mixed in water with equimolar amount of glycine, then water added to form a final 1.19 M copper in solution.

Composition G: a measured amount of zinc (supplied in the form of ZnSO<sub>4</sub>.7H<sub>2</sub>O), was thoroughly mixed in water with equimolar amount of glycine, then water added to form a final 1.6 M zinc in solution.

Composition H: an equal volume mixture of composition G and F. Tinal concentration of total metals: 1.3 M.

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## Examples

Fungal growth was determined by measuring radial size of culture growing in plates on potato dextrose agar. Test solutions at various concentrations were incorporated into plates seeded with 5 mm plug of one week old cultures. Colony diameter was determined at various times of incubation at 25°C.

Example 1. Penicillium digitatum, a common fungal pest, was treated in plates as described above, results are shown in Table 1.

Table 3c: Inhibitory Concentration of Test Compositions (in % of Culture

Medium) on Penicillium Digitatum Cultures

Inhibition marks: +++ no inhibition; ++ 40-70% inhibition;

+ 80-90 % inhibition; - total inhibition.

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Test	0	0.05%	0. 1	0.2%	0.5
Composition	%		%		%
F .	+++	++	+	-	-
G	+++	+++	++	+	-
Н	+++	++	<u> </u>		-

The results show that for each of the three above well known plant pests, no matter whether the copper chelate or the zinc chelate alone were more potent, it was the mixture of the two which was most effective.

## **Bacterial Infestation**

Following are various compositions (termed henceforth "test compositions") that were tried in various experiments, in vivo and or in vitro for their effect on bacterial pathogens. All preparatory steps were carried out accompanied by occasional stirring, until clear solution was obtained.

Composition A: 223 g citric acid, 94 g zinc oxide 450 ml water, then 50 ml nitric acid and 50 ml sulfuric acid, mixed and water added to a final volume of 700 ml.

Composition B: 223 g citric acid, 290 g cupric sulfate (CuSO<sub>4</sub>.5H<sub>2</sub>O), to that about 1200 ml water added to a final total volume of 1400 ml.

Example 3. Visual observation was used to find the inhibitory concentration of test compositions. Tissue culture wells were used to grow cultures of the fungi on agar, each consecutive well in a row containing an increasing concentration of the test composition. Assessment of inhibitory concentration of the test composition was done under the microscope.

Table 3a: Inhibitory Concentration Of Test Compositions (In % Of Culture Medium) On Botrytis Cynerea Cultures

Inhibition marks: +++ no inhibition; ++ 40-70% inhibition; + 80-90 % inhibition; - total inhibition.

Test Composition	%	0. 1 %	0.2% .	0. 5 %
F	+++	+++	++	-
G	+++	+++	+	-
Н	+++	++	•	-

Table 3b: Inhibitory Concentration of Test Compositions (in % of Culture

Medium) on *Penicillium Expansum* Cultures

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Test	0 %	0.1%	0.2%	0.5%
Composition				
F	+++	+++	+	-
G	+++	+++	++	-
Н .	+++	++	<b>-</b>	-

Inhibition marks: +++ no inhibition; ++ 40-70% inhibition;

<sup>+ 80-90 %</sup> inhibition; - total inhibition.

First spraying was performed on March 4th, at swelling of buds, further sprayings were carried out on March 12, 17, and 21; and April 14, 1996.

# List of Treatments:

- 1. 0.7% of test composition E in water with 0.025% of Triton X-100 surfactant.
- 2. 1% of the above composition, as above.
- 3. Kocide<sup>™</sup> (Copper hydroxide by Griffin Corp. USA.), 0.3% at bud swelling only;
- Starner<sup>™</sup> (Quinolinone compound, Sumitomo, Japan.), 0.15% for the rest of the five applications.
  - 5. Control (no sprays).
  - Evaluation of results: On April 26, May 7, and May 21,1996 counts were recorded of all infection centers typical to fireblight on the trees (including inflorescences, branches and fruit spurs).

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Table 4: Assessment of Fireblight Infection

Treatments	Average Number of Necrotic  Centers						
<u> </u>							
	April 26,	May 7, 1996	May 21,				
	1996		1996				
1	14.25 A	68.50 AB	97.75 AB				
2	14.00 A	35.25 A	72.25 B				
3	5.25 A	27.25 A	66.25 B				
4	31.50 A	91.75 A	134.75 A				

The statistical analysis performed was Duncan Multiple Range Test at a degree of 5% significance. In table 1, assessment figures in a column accompanied by different letters are significantly different.

Laboratory Investigation of Response of Bacterial Cultures to
5 Various Chelates

Two different bacteria strains were used throughout: one, *Erwinia* carotovora isolated from potato tuber soft rot, and *Pseudomonas spp.* isolated from soft rot of *Calla* spp tuber. Starter cultures of both strains (24 hours old) were used for inoculating liquid medium flasks each containing different concentrations of one of the test compositions described above. Growth of the bacteria in the medium was determined after 24 and 48 hours of incubation at 28°C by measuring the turbidity of the culture at 550 nm.

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Table 5a. Erwinia Growth in Culture as Affected By Test Compositions

Figures denote turbidity of culture after 48 hours for increasing concentrations of test compositions in culture flasks.

Composition					% in			
					Culture	<b>=</b>		
				· ·	e saman emeriya	· · · · · · · · · · · · · · · · · · ·	44 11 <u>11</u>	en ist op er
	Molarity	0.00	0.10	0.50	1.00	1.50	2.00	2.50
	of total						ļ	
	metals							
Α	1.4	1.10	1.10	0.00	0.00	0.00	0.00	-
C	1.13	1.21	1.14	0.00	0.00	0.00	0.00	-
F	1.19	1.05	1:05	0.40	0.00	0.00	0.00 `	-
G	1.6	1.24	1.04	0.30	0.00	0.00	0.00	-
Н	1.06	1.18	1.04	0.33	0.11	0.00	0.00	-
i	1.2	1.26	1.15	0.08	0.00	0.00	0.00	<u> </u>

1.33	1.09	0.97	0.00	0.00	0.00	0.00	
1,1							

Table 5b. *Pseudomonas* Growth in Culture as Affected By Test

Compositions

Figures denote turbidity of culture after 48 hours for increasing concentrations of test compositions in culture flasks.

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	<u> </u>	[		%	in Cult	ure		
Comp	osition		-					
	Molarity of total metals	0.00	0.10	0.50	1.00	1.50	2.00	2.50
С	1.13	1.23	0.59	0.05	0.00	0.00	0.00	-
F	1.19	1.22	1.18	0.27	0.00	0.00	0.00	<u> </u>
G	1.6	1.15	0.00	0.00	0.00	0.00	0.00	-
H	1.06	1.15	0.17	0.14	0.07	0.00	0.00	<u>  -                                   </u>
	1.2	1.13	0.93	0.15	0.00	0.00	0.00	-
	1.33	1.04	0.02	0.00	0.00	0.00	0.00	-

The results of the above laboratory investigations show activity of all test compositions. Indication for specific activity of different compositions towards each of the two strains of bacteria are clearly noted. In addition, for each strain specific composition affect the bacteria to a different extent. Attention is drawn to test composition G, that was most concentrated. However in *Erwinia* it can be seen that for 0.5% concentration, test composition I was more effective although its molar concentration was lower. Composition G was much more effective inhibitor of growth of the *Pseudomonas* strain than it was of the *Erwinia* strain.

# Field Experiment 2: Phytotoxic Considerations

New, developing leaves of pear cultivar Spadona in early spring were sprayed with either 1% solution of the test solution C in water and surfactant as

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above, whilst other trees were sprayed with test solution J in either 1% solution in water or in 0.6% concentration of the same. After three days, substantial scorching was observed in the leaves of the group sprayed with the 1% of solution J. Scorching also appeared in 0.6% of the same solution, in slightly less severe symptoms. The solution of 1% C in water exhibited almost no Phytotoxic symptoms, after the three day period and later on.

#### CLAIMS

- 1. A composition for controlling fungal diseases which contains at least one type of chelating agent complexed with zinc ion and with at least another metal ion, whereas the chelating agent molecules are characterized in being either synthetic or natural molecules.
- 2. A composition for controlling fungal diseases which contains at least one type of chelating agent complexed with zinc ion and with at least another metal ion considered to have nutritive value, whereas the chelating agent molecules are characterized in being either synthetic or natural molecules.
- 3. A composition as in Claim 1 wherein the ingredients are in the form of aqueous solution.
  - 4. A composition as in Claim 1 wherein the ingredients are in dry form, whereby the addition of measured amount of water is performed prior to application onto plants.
- 5. A method for protecting plants or parts of plants against fungal pests comprising applying to said plants a composition containing chelated zinc mixed with at least another chelated metal.
  - 6. A composition for the control of bacterial diseases which contains at least one type of metal chelated by at least one type of chelating agent, whereas the chelating agent molecules are characterized in being any one or a combination of the following:

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a) organic acid containing optionally more than one acidic group,

b) organic acid containing apart from the acidic group an additional functional group,

- c) an alcohol or an amine optionally containing more than one functional groups,
  - d) an aromatic ring substitution product,
  - e) a synthetic chelating agent.

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- 7. A composition as in Claim 6 wherein the ingredients are in the form of aqueous solution.
- 8. A composition as in Claim 6 wherein the ingredients are in dry form,
  whereby the addition of measured amount of water is performed prior to application onto plants.
  - 9. A composition according to claim 6, wherein one metal is zinc.
  - 10. A method for the control of bacterial infestations comprising:

    administration of a composition containing at least one chelated metal species of the group known to form chelates, whereas a combination of at least one type of metal with at least one type of chelating agents is used against a certain bacterial infestation, or as a precautionary measure before an infestation is expected to begin.
  - 11. A method according to Claim 10, in which a bacterial strain is controlled better by a certain combination than by another combination.
    - 12. A method according to Claim 10, in which Phytotoxic symptoms are less pronounced in using one type of mixture than in using another.

13. A method according to Claim 10, whereas at least one of the metals used alone or in combination is of a group known to be nutritious for plants or bacteria.

14. A method according to Claim 10, wherein the chelate composition is administered to plant roots, via trunks, for systemic effect, by injection, by drippers or by irrigation.

Composition C: an equal volume mixture solution of compositions A and B.

<u>Composition D</u>: 223 g citric acid, 196 g manganese sulfate (MnSO<sub>4</sub>.1H<sub>2</sub>O), to that water added to a final volume of 800 ml.

Composition E: an equal volume mixture solution of A, B and D.

Composition F: 10 g glycine, 38.6 g cupric sulfate (CuSO<sub>4</sub>.5H<sub>2</sub>O), water added to a final volume of 132 ml.

Composition G: 28 g glycine, 30 g-zinc oxide, 100 ml water, 25 ml sulfuric acid, then water added to a final volume of 235 ml.

Composition H: 80% concentration of composition E.

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Composition I: 30 g ZnO, 200 water, 30 ml sulfuric acid, 41 g resorcinol, water added to a final total volume of 340 ml.

Composition J: an equal volume mixture solution of F and G.

# Field experiment 1: Effectivity of Composition in an Orchard

Pears are very susceptible to *Erwinia amylovora* attack, causing the trees a severe disease called fireblight, the symptoms of which are blackening and death of twigs and branches, with gradual progression towards the trunk, often leading to death of whole trees and entire orchards. As of today, there are no known efficient cures for the disease. An experiment was designed to assess the efficacy of a composition according to the invention against an attack by the bacterial pathogen.

The experiment was carried out in a mature pear orchard of the variety Spadona, in northern Israel. Plot size of four trees were employed for each replication, with four random placed replications of each treatment.

# INTERNATIONAL SEARCH REPORT

International application No.

PCT/IL00/00219

	B. FIE	: 424/405, 630, 639, 641, Dig. 6; 514/492, 494, 499 to International Patent Classification (IPC) or to both national classification and IPC LDS SEARCHED	
	Minimum ( U.S. :	documentation searched (classification system followed by classification symbols) 424/405, 630, 639, 641, Dig. 6; 514/492, 494, 499	
	Document	ation searched other than minimum documentation to the extent that such documents are included	ded in the fields searched
	Electronic Please See	data base consulted during the international search (name of data base and, where practicable Continuation Sheet	search terms used)
. }-	C 100	CUMENTS CONSIDERED TO BE RELEVANT	
		Circuity of document with indication, where appropriate, of the relevant passages	Relevant to claim reo:
ŀ	X	Database CAPLUS on STN, Accession Number 1976:517976, JENO, S. Systematically active fungicides. Abstract, AT 329925, see entire document.	1-14
	Y	US 3,494,945 A (LEWIS et al.) 10 February 1970 (10.02.1970), column 1, lines 35-44, column 2, lines 44, 45, 69-72, column 3, lines 1-9, column 5, lines 20-54.	1-11, 13, 14
	Y	JP 60146808 A (RIKAGAKU KENKYUSHO) 02 August 1985 (02.08.1985), see attached Abstract translation.	1-11, 13, 14
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	Fur	ther documents are listed in the continuation of Box C. See patent family annex.	miernational filing date or priorny
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	"E" cari	er application or patent published on or after the international filling date  considered novel or cannot be considered novel	ne ciamed investing cannot be idered to involve an inventive step
	esta	ment which may throw doubts on priority claim(s) or which is cited to  blish the publication date of another citation or other special reason (as "Y"  considered to involve an inventive continued with one or more other.	siep when the document is
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INTERNATIONAL SEARCH REPORT	Internatic al application No.
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Continuation of B. FIELDS SEARCHED Item 3: STN/CAS, WEST	
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